



The Polytechnic Engineer Volume 10 (Paperback)

By Books Group

Not Avail, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****. This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1910 Excerpt: .acetate whose chemical action is essentially the same as caustic soda and dried sodium acetate heated together, thus $\text{CH}_3\text{CO} \cdot \text{ONa} + \text{NaOII} = \text{CH}_4 + \text{Na}_2\text{CO}_3$. The thoroughly dried gas was passed over the previously dried metallic oxides placed in the bend of a U-shaped Marchand tube immersed in an oil bath. The bulb of the mercury thermometer was held in position against the bend of the tube. The following table shows the temperature at which dry ammonia gas and dry methane began to take oxygen from the metallic oxides and reduce them to the metallic state: III.-- Table of Reductions with Ammonia and with Methane. Oxide. Formula. Temperature with Temperature with W 13. Crl4. Mercurous oxide Hg_2O 67C. 220C. Mercuric oxide HgO 157 200-210 Cuprous oxide Cu_2O 208 230 Oxide. Formula. Temperature with Temperature with Cupric oxide CuO 225 280 Lead suboxide Pb_2O_3 202...



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